

(19) World Intellectual Property Organization  
International Bureau



(43) International Publication Date  
5 April 2001 (05.04.2001)

PCT

(10) International Publication Number  
**WO 01/23492 A1**

(51) International Patent Classification<sup>7</sup>: C09K 5/04

THOMAS, James, Victor [CA/CA]; 247 Howe Avenue,  
Fall River, Nova Scotia B2T 1H7 (CA).

(21) International Application Number: PCT/GB00/03724

(22) International Filing Date:  
29 September 2000 (29.09.2000)

(74) Agent: BROWNE, Robin, Forsythe; Urquhart-Dykes &  
Lord, Tower House, Merrion Way, Leeds LS2 8PA (GB).

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:  
9923088.0 30 September 1999 (30.09.1999) GB  
0011653.3 15 May 2000 (15.05.2000) GB

(81) Designated States (*national*): AE, AG, AL, AM, AT, AU,  
AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CR, CU, CZ,  
DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR,  
HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR,  
LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ,  
NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM,  
TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW.

(71) Applicant (*for all designated States except US*): RE-  
FRIGERANT PRODUCTS LTD [GB/GB]; N9 Central  
Park Estate, Westinghouse Road, Trafford Park, Manches-  
ter M17 1PG (GB).

(84) Designated States (*regional*): ARIPO patent (GH, GM,  
KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian  
patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European  
patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE,  
IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG,  
CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).

(72) Inventors; and

(75) Inventors/Applicants (*for US only*): POWELL, Richard  
[GB/GB]; 9 Sadlers Wells, Bunbury, Cheshire CW6 9NW  
(GB). POOLE, John, Edward [GB/GB]; Wythorpe,  
Murieston Road, Hale, Altrincham, Cheshire WA15  
9ST (GB). CAPPER, John, Derek [GB/GB]; 45 Grove  
Mount, Davenham, Northwich, Cheshire CW9 8LY (GB).

Published:

— With international search report.

For two-letter codes and other abbreviations, refer to the "Guid-  
ance Notes on Codes and Abbreviations" appearing at the begin-  
ning of each regular issue of the PCT Gazette.

(54) Title: R 12 REPLACEMENT REFRIGERANT

(57) Abstract: A refrigerant composition comprising 1,1,1,2-tetrafluoroethane (R134a) and pentane wherein the weights of pentane and R134a are in the range: Pentane 1-5 %; R134a 99-95 %.

WO 01/23492 A1

## R 12 REPLACEMENT REFRIGERANT

This invention relates to a refrigerant particularly but not exclusively for air conditioning systems. The system relates especially to refrigerant compositions which have no adverse effect on the atmospheric ozone layer and to compositions which can be added to existing refrigerants which are compatible with lubricants commonly used in refrigeration and air conditioning systems. The invention also relates to a method of modifying refrigeration and air conditioning systems.

Chlorofluorocarbons (CFCs) eg CFC 11 and CFC 12 are stable, of low toxicity and non-flammable providing low hazard working conditions used in refrigeration and air conditioning systems. When released they permeate into the stratosphere and attack the ozone layer which protects the environment from damaging effects of ultraviolet rays. The Montreal Protocol, an International environmental agreement signed by over 160 countries, mandates the phase-out of CFCs according to an agreed timetable. This now includes hydrochlorofluorocarbons (HCFCs) which also have an adverse effect on the ozone layer.

Any replacement for R 12 must have no ability to deplete ozone. The compositions of the present invention do not include chlorine atoms and consequently they will have no deleterious effect on the ozone layer while providing a similar performance as a working fluid to R 12 in refrigeration apparatus.

Various terms have been used in patent literature to describe refrigerant mixtures. These may be defined as follows:

**Zeotrope:** A fluid mixture whose vapour and liquid compositions are different at a specified temperature.

**Temperature glide:** If a zeotropic liquid is distilled at constant pressure its boiling point will increase. The change in boiling point from the beginning of the distillation until the point when a liquid phase has just disappeared is called the temperature glide. A glide is also observed when the saturated vapour of a zeotrope is condensed at constant pressure.

**Azeotrope:** A fluid mixture of specified composition whose vapour and liquid compositions are the same at a specified temperature. Strictly speaking a fluid mixture which is an azeotrope under for example evaporator conditions, cannot also be an

-2-

azeotrope under the condenser conditions. However the refrigeration literature may describe a mixture as azeotropic provided that it meets the above definition at some temperature within its working range.

**Near-azeotropes:** A blend which boils over a small temperature range, that has a small temperature glide.

**Retrofit refrigerant mixture:** A non-chlorine-containing mixture used to replace completely the original CFC or HCFC refrigerant.

**Extender refrigerant mixture:** A non-chlorine-containing mixture added during servicing to the CFC or HCFC refrigerant remaining in a unit, that is a top up refrigerant to make good any leakage.

**Hermetic compressor:** A compressor where the electric motor is in the same totally welded casing as the compressor. The motor is cooled by the refrigerant vapour returning to the compressor. The heat generated by the motor is removed through the condenser.

**Semi-hermetic compressor:** Similar to a hermetic compressor, the major difference being the casing has a bolted joint which can be opened to enable the motor and compressor to be serviced.

**Open compressor:** A compressor which is driven by an external motor via a drive shaft passing through the compressor casing. The motor heat is dissipated directly to the environment, not via the condenser. This results in a slightly more efficient performance than a hermetic compressor, but refrigerant leaks can occur at the shaft seal.

Percentages and proportions referred to in this specification are by weight unless indicated otherwise. Percentages and proportions are selected to total 100%.

According to a first aspect of the present invention a refrigerant composition comprises 1,1,1,2-tetrafluoroethane (R 134a) and pentane wherein the weights of pentane and R 134a are in the range

Pentane	1 - 5%
R 134a	99 - 95%

The preferred weights of pentane and R134a are in the range.

Pentane	2 - 3%
R 134a	98 - 97%

-3-

Positive displacement compressors, that is reciprocating or rotary compressors, used in refrigeration systems suck in small amounts of lubricant from the crank case which are ejected with the refrigerant vapour through the exhaust valves. In order to maintain compressor lubrication this oil must be forced around the circuit by the refrigerant stream and returned to the crank case. CFC and HCFC refrigerants are miscible with hydrocarbon oils and hence carry the oils around the circuit. However HFC refrigerants and hydrocarbon lubricants have low mutual solubilities so effective oil return may not occur. The problem is particularly acute in evaporators where low temperatures can increase the viscosities of oils sufficiently to prevent them being carried along the tube walls. With CFCs and HCFCs enough refrigerant remains in the oil to reduce the viscosities to enable oil return to occur.

When using HFCs with hydrocarbon lubricants oil return can be facilitated by introducing into the system a hydrocarbon fluid having the following properties:

- (a) sufficient solubility in the lubricant at the evaporator temperature to reduce its viscosity; and
- (b) sufficient volatility to allow distillation from the hot lubricant in the compressor crank case.

Hydrocarbons fulfil these requirements.

Preferred hydrocarbons additives are selected from the group consisting of: n-pentane, cyclopentane, isopentane and mixtures thereof. Use of n-pentane, isopentane or mixtures thereof is especially preferred.

The amount of pentane may be up to 5%, preferably 1 to 5 % and more preferably about 2 - 3%.

An additional hydrocarbon may be employed for example: 2-methylpropane, 2,2-dimethylpropane, butane, pentane, 2-methylbutane, cyclopentane, hexane, 2-methylpentane, 3-methylpentane, 2,2-dimethylbutane and methylcyclopentane. Use of butane is preferred. Use of a lower boiling hydrocarbon boosts capacity and enhances oil return. The total amount of pentane and additional hydrocarbons may be selected to equal the amounts of pentane set out above.

In particularly embodiments of the invention a mixture of pentane, preferably n-pentane, isopentane or a mixture hereof together with butane is employed. This provides the advantage that a close boiling or near azeotropic blend may be obtained so that

formation of a flammable high proportion of pentane is avoided in the event of leakage, for example from a storage cylinder.

Relative proportions of the pentane and butane components may be selected to give a total of 0.2 to 5% of the composition, preferably 2 to 4%, more preferably 3 to 4%. An amount of pentane, preferably isopentane of 0.2 to 2% may be used together with a corresponding amount of 4.8 to 3% of butane in a composition containing a total of 5% hydrocarbon. In compositions with less than 5% hydrocarbon, for example 1% or 4%, relatively larger ratios of butane : pentane may be employed to minimise hydrocarbon build-up on leakage. Flammability risks are therefore reduced.

A particularly preferred compositions comprises:

R 134a	96 - 97%
--------	----------

Pentane/butane mixture	3 - 4%
------------------------	--------

A ratio of pentane/butane of 1 : 3 to 1 : 8, preferably about 1 : 5 may be employed.

Refrigerant compositions in accordance with this invention confer several advantages. The presence of pentane in the mixture increases the solubility properties of the mixture with traditional lubricants, for example mineral and alkyl benzene oils.

The present invention may confer a number of benefits in comparison to R 12 including lower intrinsic global warming potential and lower discharge temperature. The present invention may confer a number of benefits in comparison to pure R 134a including greater miscibility and higher solubility in hydrocarbon oils and hence better oil return.

The invention is further described by means of examples but not in any limitative sense.

#### EXAMPLE 1

The performance of five R134a/pentane compositions was evaluated using standard refrigeration cycle analysis techniques in order to assess their suitability as retrofit replacements for R12 in hermetic or semi-hermetic systems. The operating conditions used for the analyses were chosen as being typical of the conditions found in refrigeration systems. Since the blends were zeotropes the midpoints of their temperature glides in the evaporator and condenser were chosen to define the temperature limits of the cycle. The same temperatures were also used to generate performance data from R12.

-5-

The following refrigerant compositions were subjected to cycle analysis:

1. A composition comprising 1% pentane, 99% 134a
2. A composition comprising 2% pentane, 98% 134a
3. A composition comprising 3% pentane, 97% 134a
4. A composition comprising 4% pentane, 96% 134a
5. A composition comprising 5% pentane, 95% 134a

The following cycle conditions were used in the analysis:

COOLING DUTY DELIVERED	10 kW
EVAPORATOR	
Midpoint fluid evaporation temperature	-20°C
Superheating	5.0°C
Suction line pressure drop (in saturated temperature)	1.5°C
CONDENSER	
Midpoint fluid condensing temperature	45.0°C
Subcooling	5.0°C
Discharge line pressure drop (in saturated temperature)	1.5%
LIQUID LINE/SUCTION LINE HEAT EXCHANGER	
Efficiency	0.3%
COMPRESSOR	
Electric motor efficiency	0.85
Compressor isentropic efficiency	0.7
Compressor volumetric efficiency	0.82
PARASITIC POWER	
Evaporator fan	0.3 kW
Condenser fan	0.4 kW
Controls	0.1 kW

-6-

The results of analysing the performances in a refrigeration unit using these operating conditions are shown in Table 1. For comparison the performance of R12 is also shown. All compositions have lower discharge temperatures than R12 and are therefore superior on this account. However with higher pentane contents the cooling capacity is reduced so compositions with the minimum adequate pentane content to ensure oil return are preferred, notably those with between 2 and 3%.

TABLE 1

Refrigerant % by weight	1. 134a/pentane 99/1	2. 134a/pentane 98/2	3. 134a/pentane 97/3	4. 134a/pentane 96/4	5. 134a/pentane 95/5	R-12
Discharge pressure (bar)	11.77	11.49	11.20	10.91	10.63	11.21
Discharge temperature (°C)	118	119	119	120	121	128
COP (system)	1.34	1.34	1.34	1.34	1.33	1.36
Capacity (kW/m <sup>3</sup> )	633	613	590	566	542	698
Glide in evaporator	0.97	2.10	3.40	4.87	6.49	0
Glide in condenser (°C)	0.91	1.88	2.91	3.99	5.13	0



EXAMPLE 2

The performance of five R134a/pentane compositions was evaluated using standard refrigeration cycle analysis in order to assess their suitability as retrofit replacements for R12 in open systems. The operating conditions, used for the analyses were chosen as being typical of conditions found in refrigeration systems. Since the blends were zeotropes the midpoints of their temperature glides in the evaporator and condenser were chosen to define the temperature limits of the cycle. The same temperatures were also used to generate performance data for R12.

The following compositions were subjected to cycle analysis;

1. A composition comprising 1% pentane, 99% 134a
2. A composition comprising 2% pentane, 98% 134a
3. A composition comprising 3% pentane, 97% 134a
4. A composition comprising 4% pentane, 96% 134a
5. A composition comprising 5% pentane, 95% 134a

The following cycle conditions were used in the analysis:

COOLING DUTY	10 kW
EVAPORATOR	
Midpoint fluid evaporation temperature	-10.0°C
Superheating	5.0°C
Suction line pressure drop (in saturated temperature)	1.5°C
CONDENSER	
Midpoint fluid condensing temperature	45.0°C
Subcooling	5.0°C
Discharge line pressure drop (in saturated temperature)	1.5%

-9-

## LIQUID LINE/SUCTION LINE HEAT EXCHANGER

Efficiency	0.3%
------------	------

## COMPRESSOR

Electric motor efficiency	0.85
---------------------------	------

Compressor isentropic efficiency	0.7
----------------------------------	-----

Compressor volumetric efficiency	0.82
----------------------------------	------

## PARASITIC POWER

Evaporator fan	0.3 kW
----------------	--------

Condenser fan	0.4 kW
---------------	--------

Controls	0.1 kW
----------	--------

The results of analysing the performances in an air-conditioning unit using these operating conditions are shown in Table 2. For comparison the performance of R12 is also shown. All compositions have lower discharge temperatures than R12 and are therefore superior on this account. However with higher pentane contents the cooling capacity is reduced so compositions with the minimum adequate pentane content to ensure oil return are preferred, notably those with between 2 and 3%

TABLE 2

Refrigerant % by weight	1. 134a/pentane 99/1	2. 134a/pentane 98/2	3. 134a/pentane 97/3	4. 134a/pentane 96/4	5. 134a/pentane 95/5	R-12
Discharge pressure (bar)	11.77	11.49	11.20	10.91	10.63	11.21
Discharge temperature (°C)	89.7	90.0	90.4	90.9	91.5	94.8
COP (system)	2.544	2.54	2.54	2.54	2.53	2.57
Capacity (kW/m <sup>3</sup> )	1051	1020	987	953	917	1111
Glide in evaporator	0.97	2.08	3.32	4.69	6.20	0.00
Glide in condenser (°C)	0.91	1.88	2.91	3.99	5.13	0.00

EXAMPLE 3

The performance of five R134a/pentane compositions was evaluated using standard refrigeration cycle analysis in order to assess their suitability as retrofit replacements for R12 in mobile air conditioning systems. The operating conditions used for the analyses were chosen as being typical of conditions found in refrigeration systems. Since the blends were zeotropes the midpoints of their temperature glides in the evaporator and condenser were chosen to define the temperature limits of the cycle. The same temperatures were also used to generate performance data for R12

The following refrigerant compositions were subjected to cycle analysis:

1. A composition comprising 1% pentane, 99% 134a
2. A composition comprising 2% pentane, 98% 134a
3. A composition comprising 3% pentane, 97% 134a
4. A composition comprising 4% pentane, 96% 134a
5. A composition comprising 5% pentane, 95% 134a

The following cycle conditions were used in the analysis:

COOLING DUTY DELIVERED	10 kW
------------------------	-------

#### EVAPORATOR

Midpoint fluid evaporation temperature	7.0°C
Superheating	5.0°C
Suction line pressure drop (in saturated temperature)	1.5°C

#### CONDENSER

Midpoint fluid condensing temperature	60.0°C
Subcooling	5.0°C
Discharge line pressure drop (in saturated temperature)	1.5%

-12-

**COMPRESSOR**

Compressor isentropic efficiency	0.7
Compressor volumetric efficiency	0.82

**PARASITIC POWER**

Condenser fan	0.4 Kw
---------------	--------

The results of analysing the performances in an air-conditioning unit using these operating conditions are shown in Table 3. For comparison the performance of R12 is also shown. All compositions have lower discharge temperatures than R12 and are therefore superior on this account. However with higher pentane contents the cooling capacity is reduced so compositions with the minimum adequate content to ensure oil return are preferred, notably those with between 2 and 3%.

TABLE 3

Refrigerant % by weight	1. 134a/pentane 99/1	2. 134a/pentane 98/2	3. 134a/pentane 97/3	4. 134a/pentane 96/4	5. 134a/pentane 95/5	R-12
Discharge pressure (bar)	17.04	16.65	16.27	16.27	15.51	15.72
Discharge temperature (°C)	84.4	85.3	85.9	85.9	87.4	88.4
COP (system)	2.38	2.38	2.38	2.38	2.38	2.45
Capacity (kW/m <sup>3</sup> )	1730	1687	1643	1643	1552	1754
Glide in evaporator	0.91	1.91	3.02	3.02	5.52	0.00
Glide in condenser (°C)	0.83	1.71	2.64	2.64	4.63	0.00

**EXAMPLE 4**

The performance of two R134a/pentane/butane compositions was evaluated using standard refrigeration cycle analysis techniques in order to assess their suitability as retrofit replacements for R12 in hermetic or semi-hermetic systems. The operating conditions used for the analyses were chosen as being typical of conditions found in refrigeration systems. Since the blends were zeotropes the midpoints of their temperature glides in the evaporator and condenser were chosen to define the temperature limits of the cycle. The same temperatures were also used to generate performance data for R12.

The following refrigerant compositions were subjected to cycle analysis:

1. A composition comprising 1% pentane, 2.5% butane, 96.5% 134a
2. A composition comprising 1.7% pentane, 2.6% butane, 95.7% 134a.

The following cycle conditions were used in the analysis:

COOLING DUTY DELIVERED	10 kW
------------------------	-------

EVAPORATOR

Midpoint fluid evaporation temperature	7-20°C
Superheating	5.0°C
Suction line pressure drop (in saturated temperature)	1.5°C

CONDENSER

Midpoint fluid condensing temperature	45.0°C
Subcooling	5.0°C
Discharge line pressure drop (in saturated temperature)	1.5%

LIQUID LINE/SUCTION LINE HEAT EXCHANGER

Efficiency	0.3
------------	-----

COMPRESSOR

Electric motor efficiency	0.85
Compressor isentropic efficiency	0.7

-15-

Compressor volumetric efficiency	0.82
PARASITIC POWER	
Evaporator fan	0.3 kW
Condenser fan	0.4 kW
Controls	0.1 kW

The results of analysing the performances in a refrigeration unit using these operating conditions are shown in Table 1. For comparison the performance of R12 is also shown. All compositions have lower discharge temperatures than R12 and are therefore superior on this account. However with higher pentane contents the cooling capacity is reduced so compositions with the minimum adequate pentane content to ensure oil return are preferred, notably those with between 2 and 3%



TABLE 4

Refrigerant % by weight	Pentane 1.0 butane 2.5 134a 96.5	Pentane 1.0 butane 2.6 134a 95.7	R-12
Discharge pressure (bar)	11.24	11.04	11.21
Discharge temperature (°C)	118	119	128
COP (system)	1.34	1.34	1.36
Capacity (kW/m <sup>3</sup> )	631	616	698
Glide in evaporator (°C)	1.09	1.96	0
Glide in condenser (°C)	1.03	1.77	0

EXAMPLE 5

The performance of two R134a/pentane/isobutane compositions were evaluated using standard refrigeration cycle analysis techniques in order to assess their suitability as retrofit replacements for R12 in hermetic or semi-hermetic systems. The operating conditions used for the analyses were chosen as being typical of conditions found in refrigeration systems. Since the blends were zeotropes the midpoints of their temperature glides in the evaporator and condenser were chosen to define the temperature limits of the cycle. The same temperatures were also used to generate performance data for R12.

The following refrigerant compositions were subjected to cycle analysis:

3. A composition comprising 1.7% pentane, 1.9% isobutane, 96.4% 134a
4. A composition comprising 1.7% pentane, 1.1% isobutane, 97.2% 134a

The following cycle conditions were used in the analysis:

COOLING DUTY DELIVERED	10 kW
------------------------	-------

#### EVAPORATOR

Midpoint fluid evaporation temperature	7-20°C
Superheating	5.0°C
Suction line pressure drop (in saturated temperature)	1.5°C

#### CONDENSER

Midpoint fluid condensing temperature	45.0°C
Subcooling	5.0°C
Discharge line pressure drop (in saturated temperature)	1.5%

#### LIQUID LINE/SUCTION LINE HEAT EXCHANGER

Efficiency	0.3
------------	-----

#### COMPRESSOR

Electric motor efficiency	0.85
Compressor isentropic efficiency	0.7
Compressor volumetric efficiency	0.82

-18-

## PARASITIC POWER

Evaporator fan	0.3 kW
Condenser fan	0.4 kW
Controls	0.1 kW

The results of analysing the performances in a refrigeration unit using these operating conditions are shown in Table 1. For comparison the performance of R12 is also shown. All compositions have lower discharge temperatures than R12 and are therefore superior on this account.

TABLE 5

Refrigerant % by weight	Pentane 1.7 isobutane 1.9 134a 96.4	Pentane 1.7 isobutane 1.1 134a 97.2	R-12
Discharge pressure (bar)	11.73	11.67	11.21
Discharge temperature (°C)	118	118	128
COP (system)	1.34	1.34	1.36
Capacity (kW/m <sup>3</sup> )	628	624	698
Glide in evaporator (°C)	2.13	1.98	0.00
Glide in condenser (°C)	1.83	1.74	0.00

## CLAIMS

1. A refrigerant composition comprising 1,1,1,2-tetrafluoroethane (R134a) and pentane wherein the weights of pentane and R134a are in the range:

Pentane	1 - 5%
R134a	99 - 95%

2. A refrigerant composition as claimed in claim 1, wherein the weights of pentane and R134a are in the range:

Pentane	2 - 3%
R134a	98 - 97%

3. A refrigerant composition as claimed in claim 1 or 2, further comprising an additional hydrocarbon wherein the combined weight of pentane plus additional hydrocarbon is in the range 1 - 5%.

4. A refrigerant composition as claimed in claim 3, wherein the combined weight of pentane and additional hydrocarbon is in the range 3 - 4%.

5. A refrigerant composition as claimed in claim 3, wherein the combined weight of pentane and additional hydrocarbon is in the range 2 - 3%.

6. A refrigerant composition as claimed in any of claims 3 to 5, wherein the additional hydrocarbon is butane.

# INTERNATIONAL SEARCH REPORT

Intern. Application No

PCT/GB 00/03724

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C09K5/04

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C09K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, PAJ, WPI Data

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 97 16501 A (ICI PLC ;MORRISON JAMES DAVID (GB)) 9 May 1997 (1997-05-09) claims 1-11	1,2
Y	the whole document	3-6
X	WO 94 18282 A (DU PONT) 18 August 1994 (1994-08-18) page 7, line 2 - line 18; claims 1,6; example 2	1,2
Y		3-6



Further documents are listed in the continuation of box C.



Patent family members are listed in annex.

### \* Special categories of cited documents :

- \*A\* document defining the general state of the art which is not considered to be of particular relevance
- \*E\* earlier document but published on or after the international filing date
- \*L\* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- \*O\* document referring to an oral disclosure, use, exhibition or other means
- \*P\* document published prior to the international filing date but later than the priority date claimed

- \*T\* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- \*X\* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- \*Y\* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- \*G\* document member of the same patent family

Date of the actual completion of the international search

24 November 2000

Date of mailing of the international search report

01/12/2000

Name and mailing address of the ISA

European Patent Office, P.B. 5618 Patentlaan 2  
NL - 2280 HV Rijswijk  
Tel (+31-70) 340-2040, Tx. 31 651 epo nl,  
Fax (+31-70) 340-3016

Authorized officer

olde Scheper, B

# INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/GB 00/03724

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 9716501 A	09-05-1997	AU 7314996 A BR 9611201 A CA 2236273 A EP 0858490 A JP 11514640 T	22-05-1997 06-04-1999 09-05-1997 19-08-1998 14-12-1999
WO 9418282 A	18-08-1994	US 5458798 A AU 6096594 A CN 1119454 A CN 1174970 A EP 1028152 A EP 0682683 A JP 8506581 T US 5624596 A US 5635098 A US 5670079 A	17-10-1995 29-08-1994 27-03-1996 04-03-1998 16-08-2000 22-11-1995 16-07-1996 29-04-1997 03-06-1997 23-09-1997